

COMPOSTING RESEARCH REVIEW

This document was prepared to provide a compilation of data from world-wide zeolite studies to be used as an informational resource.

WHAT IS COMPOSTING?

Composting of waste is an aerobic (requires oxygen) natural process of “rotting” or decomposing organic matter by microorganisms under controlled conditions. The composting process involves four main components: organic matter, moisture, oxygen, and bacteria.

WHY COMPOST?

- To convert organically bound nitrogen to a form of nitrogen to a usable form of nitrogen
- Kill pathogens, weeds seeds and larvae
- Reduce the volume of material
- Provide a soil amendment
- Reduce contamination of groundwater

BENEFITS OF ZEOLITE IN COMPOST

REDUCES ODOR AND GREENHOUSE GAS EMISSIONS

- Zeolite captures ammonium that is the source of the ammonia gas that is the aerosol of odors.

INCREASES THE VALUE OF COMPOST

- Zeolite retains the nitrogen in the form of ammonium.
- The Zeolite also contains more than 3% potassium.

REDUCES VOLUME

- Reduces the volume of compost feedstock by up to 50%.

REDUCES FLIES*

- By absorbing ammonia gas.

HOLDS NITROGEN IN GROWTH ZONE

- Zeolite holds nitrogen in the growth zone where it is not water soluble, but plant accessible on an as needed basis.

INCREASES CROP YIELD

- Increases seed germination and growth.
- 20% higher yield.

IMPROVES SOIL QUALITY

- Increases nutrients, porosity, oxygen content, and mediates the pH.

HOLDS MOISTURE IN GROWTH ZONE

- Holds up to 50% of its weight in water and rehydrates at night when it is cooler.

PROTECTS GROUNDWATER FROM CONTAMINATION

- Inhibits the oxidation of ammonium to nitrates that are very water soluble and contaminate the groundwater.



COMPOST TURNER

IMPROVES MICROORGANISM DEVELOPMENT

- Counteracts the inhibitory effect of ammonia overloads on microorganisms.
- Microorganisms colonize on the surfaces of Zeolite where nutrients and water are available.

MANAGES OXYGEN LEVELS

- Zeolite porosity maintains the space needed for oxygen.

MEDIATES pH

- Maintains pH levels for composting and nitrogen utilization in soil.

MAINTAINS TEMPERATURE

- Enabled aeration for metabolic heat generation by aerobic microorganisms.
- Helps maintain the heat required to kill weed seeds, pathogens, and fly larvae.

MANAGES WATER

- Zeolite holds more than 50% of its weight in water in open channel-ways and reduces evaporation from the compost.

REDUCES TIME

- Decreases the developmental lag phase of microorganisms.



Ammonia & Steam

HOW TO APPLY ZEOLITE

TOP DRESS FRESH MANURE AND COMPOST

- Captures ammonium before it can convert to ammonia gas (NH_3), reducing nitrogen losses by 50%.
- Reduces odor and fly attraction.
- Top Dress Rate: 1 to 2%

NITROGEN FORMS

Organically bound nitrogen

Bound in organic matter and unavailable to plants

Ammonia (NH_3)

Gas

Ammonium (NH_4^+)

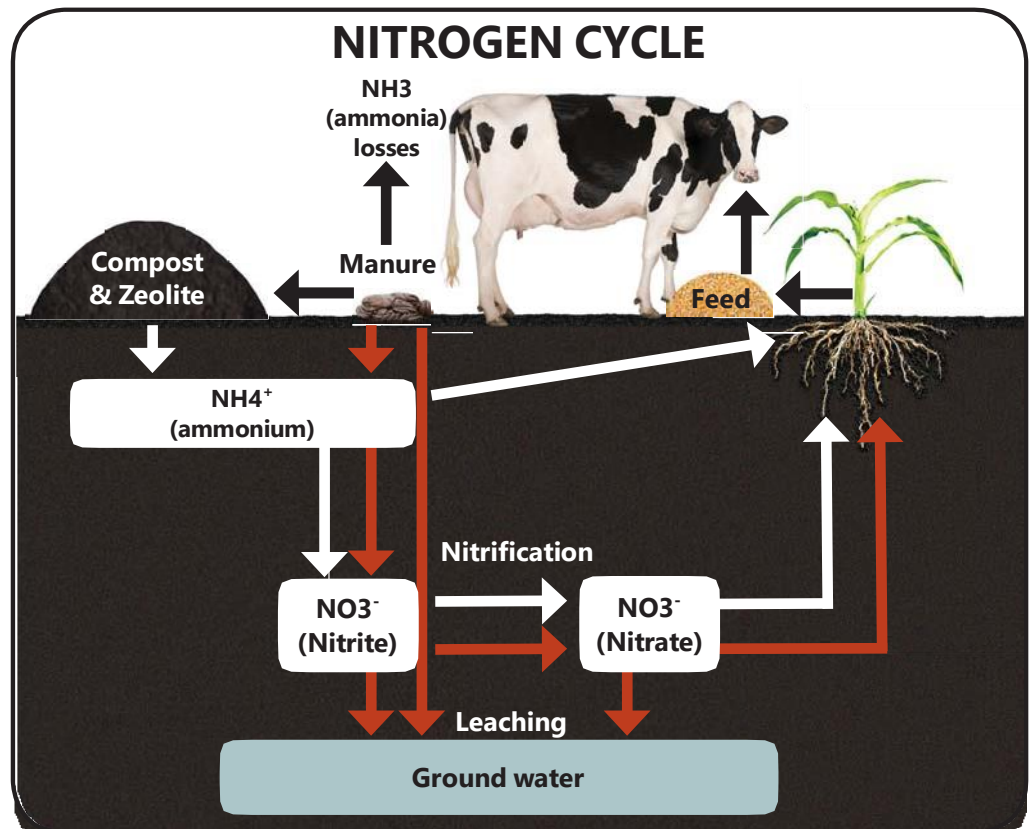
Plant available

Nitrite (NO_2^-)

Oxidizing bacteria convert ammonia to nitrite, which is toxic to plants

Nitrate (NO_3^-)

Nitrite-oxidizing bacteria convert nitrite to nitrate, a plant available nitrogen, which is very water soluble



COMPOSTING PARAMETERS

COMPOST COMPONENTS: consist of (1) nutrients, (2) carbon and nitrogen

Nutrients: Adequate phosphorous, potassium, calcium, iron, boron, copper, etc. are necessary for microbial metabolism, but they are normally in the compost feedstock.

Carbon and nitrogen: The two basic ingredients in composting are carbon and nitrogen. Ideally the C:N ratio by weight should be 30:1. Too low a carbon ratio will result in excess nitrogen that is lost as ammonia gas that creates odor problems. Too high a carbon ratio inhibits the growth of microorganism populations and the compost will remain cool and the composting degradation will be slowed. When the compost is finished, the C:N ratio will be 10-15:1 because the microorganisms will convert two thirds of the carbon to carbon dioxide. The sources of carbon (generally brown in color) and nitrogen (generally green) are as follows:



Carbon

Carbon (brown)	C:N ratio
Straw	40-100:1
Corn Stalks	60:1
Bark	100-130:1
Paper	150-200:1
Wood chips and sawdust	100-500:1

Nitrogen (green)	C:N ratio
Sewage sludge (digested)	17:1
Cow manure	20:1
Poultry manure (with litter)	13-18:1
Pig manure solids	15-25:1
Horse manure	25:1



Nitrogen

MICROORGANISMS: The main microorganisms in composting are aerobic bacteria and fungi. If not enough oxygen is provided, the compost turns anaerobic (no oxygen), microorganisms die and the compost generates putrid smells, including hydrogen sulfide.

COMPOST CONDITIONS

OXYGEN: Aerobic microorganisms can survive on 5% oxygen, but >10% is considered optimal. Oxygen is most commonly provided by turning the compost or by compressed air.

MOISTURE: Composting materials should contain between 40-60% moisture.

pH: A pH of 5.5-8.5 is ideal for the microorganisms. In early composting acids tend to accumulate and this promotes the fungi and the consequent breakdown of lignin and cellulose.

TEMPERATURE: The temperature in active compost piles range from 55-70° C which destroys pathogens and weed seeds. Temperature is controlled by turning and the addition of water.

COMPOST MANAGEMENT

TURNING: If the temperature falls below 50°C or rises above 65°C, the compost is turned.

WATER: Additional water can be added during turning if the compost is too dry.

TIME: The composting process occurs within 6-9 months. However, some in-vessel operations take only 30 days.

VOLUME: The loss of carbon dioxide and water may reduce the final volume of the compost by 50% or more.

ODOR TROUBLESHOOTING

ODOR	CAUSE	SOLUTION
Ammonia	Ammonia losses are a result of low C:N ratios. NH ₃ is at equilibrium with NH ₄ at a pH of 9. At a pH of >9, the ammonium gases to ammonia. Little ammonia is generated at acidic pHs.	Add carbon
Hydrogen sulfide	Hydrogen sulfide odors are generated if the compost becomes anaerobic. Hydrogen sulfide is more difficult to disperse because hydrogen sulfide is heavier than air, and they tend to accumulate in the compost area.	Add oxygen Turn compost to aerate

METHODS

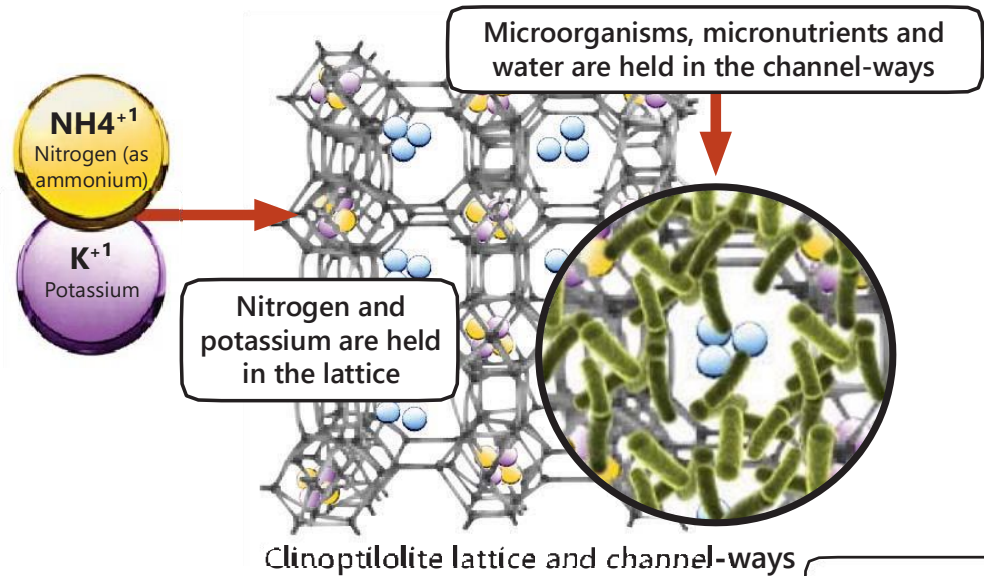
WINDROW COMPOSTING: This is common in fields and is better on a concrete or impermeable surface. The compost can be turned with tractors or compost turners.

IN-VESSEL COMPOSTING: This refers to composting in metal or plastic tanks or concrete bunkers that confine the material in buildings, containers, or vessels which protect groundwater and confine odors. These systems start with anaerobic digestion and finish with aerobic digestion.

How Zeolite Works

Zeolite has the ability to exchange ammonium (NH_4^{+1}) into its lattice through its cation exchange capacity (CEC).*

The Zeolite lattices are negatively charged and are able to hold positively charged ammonium (NH_4^{+1}) and potassium (K^{+1}), which are accessible to microorganisms as needed for growth but not water soluble.



HOW ZEOLITE COMPOST WORKS IN THE SOIL*

Zeolite contains approximately 3.47% potassium, which is an important nutrient in fertilizers. Zeolite holds at least 55% of its weight in water that protects the plant against drought.

The plant releases hydrogen (H^{+1}) during growth, which exchanges with ammonium (NH_4^{+1}) held in the Zeolite lattice, which is plant accessible but not water soluble.

Available water (H_2O) is held in the open pore spaces of the Zeolite in the growth zone.

